

Generalized Reaction-Diffusion: a microscopic approach

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In systems where diffusing particles are subject to a reactive process, diffusion and reaction are coupled and reaction-diffusion (R-D) phenomena are described at the macroscopic level by R-D equations. For instance, for an annihilation process such as $A \rightarrow 0$, the classical R-D equation reads

$$\frac{\partial}{\partial t} f(r; t) = D \frac{\partial^2}{\partial r^2} f(r; t) - k f(r; t), \quad (1)$$

where D denotes the diffusion coefficient and k the reactive rate. The classical equation (1) yields a steady state solution showing exponential decay in space $f(r) = f(0) \exp\left(-\sqrt{\frac{k}{D}} |r|\right)$.

However in most natural systems where it seems logical to use the language of reaction-diffusion, non-classical distributions are observed where the steady state shows non-exponential behavior e.g. when the particles encounter obstacles in the medium or because the reactive process is hindered or enhanced by concentration effects. So a general description of R-D phenomena requires a generalization for both diffusion and reaction.

We develop a microscopic approach by generalizing Einstein's master equation with a reactive term and we show how the mean field formulation leads to nonlinear R-D equations with non-classical solutions. For the n -th order annihilation reaction $A + A + A + \dots + A \rightarrow 0$, the generalized reaction-diffusion equation (with no drift) reads

$$\frac{\partial}{\partial t} f(r; t) = \frac{\partial}{\partial r} D \frac{\partial}{\partial r} f^\alpha(r; t) - k f^n(r; t), \quad (2)$$

giving typical solutions $f(r) = f(0) (1 + C_{\alpha, n}(D, k) r)^{\frac{-2}{n-\alpha}}$ with long range power law behavior showing the relative dominance of sub-diffusion over reaction effects or conversely leading to finite support because diffusion is slow and extinction is fast. Examples of morphogen gradient formation in biological systems are discussed.¹

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